

SolarTech Power Solutions

Lead-acid battery energy storage readable voltage



Overview

Are lead-acid batteries a good choice for energy storage?

Lead-acid batteries have been used for energy storage in utility applications for many years but it has only been in recent years that the demand for battery energy storage has increased.

What is a lead-acid battery system?

1. Technical description A lead-acid battery system is an energy storage system based on electrochemical charge/discharge reactions that occur between a positive electrode that contains lead dioxide (PbO_2) and a negative electrode that contains spongy lead (Pb).

What is a lead acid battery?

Lead-acid batteries may be flooded or sealed valve-regulated (VRLA) types and the grids may be in the form of flat pasted plates or tubular plates. The various constructions have different technical performance and can be adapted to particular duty cycles. Batteries with tubular plates offer long deep cycle lives.

Does stationary energy storage make a difference in lead-acid batteries?

Currently, stationary energy-storage only accounts for a tiny fraction of the total sales of lead-acid batteries. Indeed the total installed capacity for stationary applications of lead-acid in 2010 (35 MW) was dwarfed by the installed capacity of sodium-sulfur batteries (315 MW), see Figure 13.13.

What is the voltage of a lead-acid cell?

The voltage of a typical single lead-acid cell is ~ 2 V. As the battery discharges, lead sulfate (PbSO_4) is deposited on each electrode, reducing the area available for the reactions. Near the fully discharged state (see Figure 3), cell voltage drops, and internal resistance increases.

Why is electrochemical energy storage in batteries attractive?

Electrochemical energy storage in batteries is attractive because it is compact, easy to deploy, economical and provides virtually instant response both to input from the battery and output from the network to the battery.

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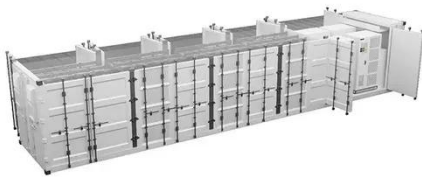


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